

# Convection (1)

181

**Convection** is the mass motion of gas elements – only occurs when temperature gradient exceeds some critical value. We can derive an expression for this.

Consider a convective element at distance  $r$  from the centre of star. Element is in equilibrium with the surrounding.

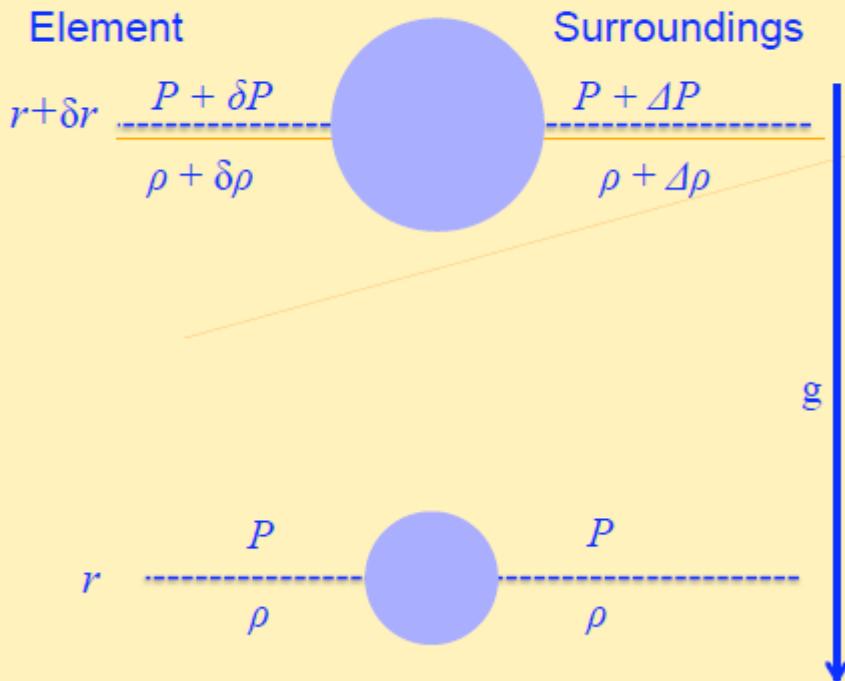
Now let's suppose it rises to  $r + \delta r$ . Element expands to stay in pressure balance with the new environment,  $P(r)$  and  $\rho(r)$  are reduced to  $P + \delta P$  and  $\rho + \delta \rho$ .

But these may not generally equal the new surrounding gas conditions.

Define those as  $P + \Delta P$  and  $\rho + \Delta \rho$ .

If gas element is denser than surroundings at  $r + \delta r$  then will sink (i.e. stable). If it is less dense then it will keep on rising – **convectively unstable**.

Convective element of stellar material



# Convection (2)

182

The condition for instability is therefore

$$\boxed{\rho + \delta\rho < \rho + \Delta\rho}$$

Whether or not this condition is satisfied depends on two things:

- The rate at which the element expands due to decreasing pressure
- The rate at which the density of the surroundings decreases with height

Let's make two assumptions

1. The element rises **adiabatically**, i.e. no heat is exchanged with the surrounding;
2. The element rises at a speed much **less than the sound speed**.

During motion, sound waves have time to smooth out the pressure differences between the element and the surroundings. Hence  $\delta P = \Delta P$  at all times.

The first assumption means that the element must obey the adiabatic relation between pressure and volume

$$PV^\gamma = \text{constant}$$

where  $\gamma = c_p / c_v$  is the **adiabatic index** or **heat capacity ratio** defined as specific heat (i.e. the energy to raise temperature of 1 g of material by 1K) at constant pressure, divided by specific heat at constant volume.

# Convection (3)

183

Given that  $V$  is inversely proportional to  $\rho$ , we can write

$$\frac{P}{\rho^\gamma} = \text{constant}$$

Hence equating the term at  $r$  and  $r + \delta r$ :

$$\frac{P + \delta P}{(\rho + \delta \rho)^\gamma} = \frac{P}{\rho^\gamma}$$

If  $\delta \rho$  is small, we can expand  $(\rho + \delta \rho)^\gamma$  using the binomial theorem as follows  
 $(\rho + \delta \rho)^\gamma \sim \rho^\gamma + \gamma \delta \rho \rho^{\gamma-1}$ . Combining last two expressions we obtain

$$\delta \rho = \frac{\rho}{\gamma P} \delta P$$

Now we need to evaluate the change in density of the surroundings,  $\Delta \rho$ .

Let's consider a very small rise of  $\delta r$

$$\Delta \rho = \frac{d \rho}{d r} \delta r$$

# Convection (4)

184

$$\rho + \delta\rho < \rho + \Delta\rho$$

And substituting these expressions for  $\delta\rho$  and  $\Delta\rho$  into the condition for convective instability derived above:

$$\delta\rho = \frac{\rho}{\gamma P} \delta P$$

$$\frac{\rho}{\gamma P} \delta P < \frac{d\rho}{dr} \delta r$$

$$\Delta\rho = \frac{d\rho}{dr} \delta r$$

And this can be rewritten by recalling our **2<sup>nd</sup> assumption** that element will remain at the **same pressure as its surroundings**, so that in the limit

$$\delta r \rightarrow 0, \quad \frac{\delta P}{\delta r} = \frac{dP}{dr}$$

$$\frac{\rho}{\gamma P} \frac{dP}{dr} < \frac{d\rho}{dr}$$

The LHS is the density gradient that **would** exist in the surroundings if they had an **adiabatic** relation between density and pressure. RHS is the **actual** density in the surroundings.

We can convert this to a more useful expression, by first dividing both sides by  $dP/dr$ . Note that  $dP/dr$  is negative, hence the **inequality sign must change**.

# Convection (5)

185

$$\frac{\rho}{\gamma P} > \frac{d\rho}{dr} / \frac{dP}{dr} \Rightarrow \frac{\rho}{\gamma P} > \frac{d\rho}{dP}$$

$$\left(\frac{P}{\rho}\right) \frac{dp}{dP} < \frac{1}{\gamma} \quad \text{or} \quad \frac{d \ln \rho}{d \ln P} < \frac{1}{\gamma}$$

For an ideal gas in which radiation pressure is negligible (where  $\mu$  is the mean molecular weight of particles in the stellar material in unit of proton mass  $m_p$ )

$$P = \frac{\rho k T}{\mu m_p} \Rightarrow \ln P = \ln \rho + \ln T + \text{const}$$

And can differentiate to give

$$\frac{dP}{P} = \frac{d\rho}{\rho} + \frac{dT}{T} \quad \text{or} \quad 1 = \frac{d \ln \rho}{d \ln P} + \frac{d \ln T}{d \ln P}$$

And combining this with the equation above gives ....

# Schwarzschild condition for occurrence of convection

186

$$\frac{P}{T} \frac{dT}{dP} = \frac{d \ln T}{d \ln P} > \frac{\gamma - 1}{\gamma}$$

which is the **Schwarzschild** condition for the occurrence of convection (in terms of the temperature gradient).

A gas is convectively unstable if the actual temperature gradient is steeper than the adiabatic gradient. If the condition is satisfied, then large scale rising and falling motions transport energy upwards.

# Condition for occurrence of convection

187

A gas is convectively unstable if **the actual temperature gradient is steeper than the adiabatic gradient**. The criterion can be satisfied in two ways:

1. The temperature gradient is very steep

For example, if a large amount of energy is released at the **centre of a star**, it may require a large temperature gradient to carry the energy away. Hence where nuclear energy is being released, convection may occur.

2. The ratio of specific heats  $\gamma$  is close to unity

Alternatively, in the cool outer layers of a star, gas may only be **partially ionized**, hence much of the heat used to raise the temperature of the gas goes into ionization and hence the specific heat of the gas at constant  $V$  is nearly the same as the specific heat at constant  $P$  (because  $T \sim \text{const}$ ), and  $\gamma \sim 1$ .

In such a case, a star can have a **cool outer convective layer**. We will come back to the issues of convective cores and convective outer envelopes later.

# Condition for occurrence of convection

188

Convection is an extremely complicated subject, and it is true to say that the lack of a good theory of convection is one of the worst defects in our present studies of stellar structure and evolution.

We know the conditions under which convection is likely to occur but don't know how much energy is carried by convection.

Fortunately, we will see that we can often find occasions where we can manage without this knowledge.

# Influence of convection

189

Let's back to the equations of stellar structure.

Ideally, we would like to know exactly **how much energy is transported by convection** – but lack of a good theory makes it difficult to predict exactly. Fully self-consistent models of stellar convection are an active area of research and require considerable computational resources to accurately capture the three-dimensional fluid dynamics.

However, it can be shown that **even a very small difference between the actual temperature gradient and adiabatic gradient is sufficient to carry all energy**. This suggests that the actual gradient is not greatly in excess of the adiabatic gradient. We can assume that the temperature gradient has exactly the adiabatic value in a convective region in the interior of a star and hence can rewrite the condition of occurrence of convection in the form

$$\frac{d \ln T}{d \ln P} = \frac{\gamma - 1}{\gamma}$$

Thus, the simplest model of convection is to assume that the process is highly efficient – so much so that it drives the system to saturate the Schwarzschild criterion.

# Equations of stellar structure in a convective region

190

Thus, **in a convective region**, we must solve the four differential equations, together with equations for  $\varepsilon$  and  $P$ :

The equation for luminosity due to radiative transport is still true:

$$L_{\text{rad}} = -\frac{256 \times \pi^2 \sigma_{\text{SB}} r^4 T^3}{3\kappa_R} \frac{dT}{dm}$$

- $\frac{dr}{dm} = \frac{1}{4\pi r^2 \rho}$
- $\frac{dP}{dm} = -\frac{Gm}{4\pi r^4}$
- $\frac{dL}{dm} = \varepsilon$
- $\frac{P}{T} \frac{dT}{dP} = \frac{\gamma-1}{\gamma}$

And once the other equations have been solved,  $L_{\text{rad}}$  can be calculated. This can be compared with  $L$  (from  $dL/dm = \varepsilon$ ) and the difference gives the value of luminosity due to convective transport  $L_{\text{conv}} = L - L_{\text{rad}}$

In solving the equations of stellar structure, the equations appropriate to a convective region must be switched on whenever the temperature gradient reaches the adiabatic value, and switched off when all energy can be transported by radiation.

# Summary

191

- We have derived the 4th equation to describe the stellar structure and explored the ways to solve these equations.
- As they are not time dependent, we must iterate with the calculation of changing chemical composition to determine short steps in the lifetime of stars. The crucial changing parameter is the **H/He** content of the stellar core (and afterwards, **He** burning will become important – to be explored in next lectures).
- We have discussed the boundary conditions applicable to the solution of the equations and made approximations, that do work with real models.
- We have also derived the condition for **convection** and explored the influence of convection on energy transport within stars. We have shown that it must be considered, but only in areas where the temperature gradient approaches the adiabatic value. In other areas, the energy can be transported by radiation alone and convection is not required. We saw that convection may be important in hot stellar cores and cool outer envelopes, but that a good quantitative theory is lacking.
- The next lectures will explore stellar interiors and the nuclear reactions.

# The equations of stellar structure - II

192

EQUATION OF STATE (EOS)  
STELLAR OPACITY

# Introduction

193

- We have 4 differential equations of stellar structure.
- Accurate expressions for pressure, opacity and energy generation are extremely complicated, but we can find simple approximate forms.
- Equations of stellar structure too complicated to find exact analytical solution, hence must be solved with computer.
- Sometimes simplifications can be made to find analytical solutions that still have most of the physics.

# The equations of stellar structure

194

- $\frac{dm}{dr} = 4\pi r^2 \rho(r)$
- $\frac{dP(r)}{dr} = -\frac{Gm}{r^2} \rho(r)$
- $\frac{dL(r)}{dr} = 4\pi r^2 \rho(r) \varepsilon(r)$
- $\frac{dT(r)}{dr} = -\frac{3}{64\pi\sigma r^2} \frac{\rho(r)\kappa_R(r)}{T^3(r)} L(r)$
- $\frac{P}{T} \frac{dT}{dP} = \frac{\gamma-1}{\gamma}$

- $r$  = radius
- $P$  = pressure at  $r$
- $m$  = mass of material within  $r$
- $\rho$  = density at  $r$
- $L$  = luminosity at  $r$  (rate of energy flow across sphere of radius  $r$ )
- $T$  = temperature at  $r$
- $\kappa_R$  = Rosseland mean opacity at  $r$
- $\varepsilon$  = energy release per unit mass per unit time

To these four differential equations we **need to add** three equations connecting the pressure, the opacity, and the energy production rate of the gas with its density, temperature, and composition:

$P = P(\rho, T, \text{chemical composition})$  → **usually called the equation of state (EOS)**  
 $\kappa_R = \kappa_R(\rho, T, \text{chemical composition})$   
 $\varepsilon = \varepsilon(\rho, T, \text{chemical composition})$

# The equation of state (EOS)

195

- The equation of state (EOS) describes the microscopic properties of stellar matter for given density  $\rho$ , temperature  $T$  and composition  $X_i$ .
- It is usually expressed as the function that relates the pressure  $P$  to  $\rho$ ,  $T$ , and mean molecular weight  $\mu$  at any place in the star.
- Since it is a solely an internal property of the gas, it can, in principle, be computed once externally, and used via a lookup table, i.e.,  $P_{\text{gas}} = P(\rho, \mu, T)$ .

# EOS in stars

196

- We have seen that stellar gas is ionized plasma, and although density is so high that typical inter-particle spacing is of the order of an atomic radius, the effective particle size is more like a nuclear radius ( $10^5$ ) times smaller.
- Thus, interior of a star contains a mixture of ions, electrons, and radiation (photons). For most stars (except for very low mass stars and stellar remnants), the ions and electrons can be treated as **an ideal gas** and quantum effects can be neglected.
- The net pressure can be divided into three components, pressure from ions, pressure from electrons, and pressure from radiation.

Total pressure:  $P = P_i + P_e + P_{\text{rad}} = P_{\text{gas}} + P_{\text{rad}}$

$P_i$  is the pressure of the ions

$P_e$  is the electron pressure

$P_{\text{rad}}$  is the radiation pressure

However,  $P_{\text{gas}}$  may **not** obey the ideal gas law due to the effects of degeneracy.

# EOS of an ideal gas

197

The equation of state for an ideal gas is:

$$P_{\text{gas}} = nkT$$

where  $n$  is concentration (number of particles per  $\text{cm}^3 = n_I + n_e$ , where  $n_I$  and  $n_e$  are the number densities of ions and electrons respectively),  $T$  is the temperature,  $k$  is Boltzmann's constant.

But we want this equation in the form:  $P = P(\rho, T, \text{chemical composition})$   
This can be written as:

$$P_{\text{gas}} = \frac{\rho kT}{\mu m_p} = \frac{\mathfrak{R} \rho T}{\mu} \quad \text{where} \quad \mathfrak{R} = \frac{k}{m_p} \quad \text{is the gas constant, and}$$

$\mu$  = mean molecular weight,  
i.e. the average mass of particles in unit of proton mass  $m_p$ .

# Mean molecular weight (1)

198

The mean molecular weight  $\mu$  (the average mass of particles in unit of proton mass  $m_p$ ) depends upon the composition of the gas and the state of ionization. For example:

- Neutral hydrogen:  $\mu = 1$
- Fully ionized hydrogen:  $\mu = 0.5$

An exact solution is **complex**, depending on fractional ionization of all the elements in all parts of the star.

For simplicity, let's now assume that all of the material in the star is fully ionized. This is justified as **hydrogen** and **helium** are most abundant and they are certainly fully ionized in stellar interiors (however, this **assumption will break down** near stellar surface).

# Mean molecular weight (2)

199

Denote abundances of different elements per unit mass by:

$X$  = fraction of material by mass of H

$Y$  = fraction of material by mass of He

$Z$  = fraction of material by mass of all heavier elements (“metals”)

$$X + Y + Z = 1$$

Hence in  $1 \text{ cm}^3$  of stellar gas of density  $\rho$ , there is mass  $X \times (\rho \text{ of H})$ ,  $Y \times (\rho \text{ of He})$ ,  $Z \times (\rho \text{ of metals})$ . In a fully ionized gas,

H gives 2 particles per  $m_H$

He gives 3/4 particles per  $m_H$  ( $\alpha$  particle, plus two  $e^-$ )

Metals, average mass  $Am_H$ , give  $\sim 1/2$  particles per  $m_H$

( $^{12}\text{C}$  has nucleus plus  $6e^- = 7/12$ )

( $^{16}\text{O}$  has nucleus plus  $8e^- = 9/16$ )

where  $A$  is the atomic weight of the species.

# Mean molecular weight (3)

200

If the density of the plasma is  $\rho$ , then add up number densities of hydrogen, helium, and metal nuclei, plus electrons from each species:

	H	He	metals
Number density of nuclei	$\frac{X\rho}{m_H}$	$\frac{Y\rho}{4m_H}$	$\frac{Z\rho}{Am_H}$
Number density of electrons	$\frac{X\rho}{m_H}$	$\frac{2Y\rho}{4m_H}$	$\frac{A}{2} \times \frac{Z\rho}{Am_H}$

The total number of particles per  $\text{cm}^3$  is then given by the sum:

$$n = 2 \frac{X\rho}{m_H} + \frac{3}{4} \frac{Y\rho}{m_H} + \frac{2Z\rho + AZ\rho}{2Am_H} \approx \frac{\rho}{m_H} \left[ 2X + \frac{3}{4}Y + \frac{1}{2}Z \right] = \frac{\rho}{\mu m_H}$$

...assuming that  $A \gg 1$

Thus,

$$\mu = \left[ 2X + \frac{3}{4}Y + \frac{1}{2}Z \right]^{-1}$$

# Mean molecular weight (4)

201

$$\mu = \left[ 2X + \frac{3}{4}Y + \frac{1}{2}Z \right]^{-1}$$

This is a good approximation to  $\mu$  except in cool, outer stellar regions.

For solar abundances,  $X = 0.73$ ,  $Y = 0.25$ ,  $Z = 0.02$ , and therefore  $\mu = 0.60$ , i.e. the mean mass of particles in a star of solar composition is a little over half the mass of the proton.

In the **central** regions of the Sun, about half of the hydrogen has already been converted into helium by nuclear reactions, and as a result  $X = 0.34$ ,  $Y = 0.64$ , and  $Z = 0.02$ , giving  $\mu = 0.85$ .

When  $Z$  is negligible:  $Y = 1 - X$ ;  $\mu = 4/(3 + 5X)$

The electron number density  $n_e$  plays a considerable role for the properties of the gas. It is convenient to introduce the mean molecular weight per electron,  $\mu_e$ , such that

$$n_e = \frac{\rho}{\mu_e m_H} \quad \Rightarrow \quad \mu_e \approx \frac{2}{1 + X}$$

**Prove it!**

# The Ionization Fraction

202

- The accurate calculation of mean molecular weight  $\mu$  requires knowledge of the chemical composition of the material and the ionization fraction. To calculate ionization fraction, one needs the Saha equation, which we will derive later, in the Stellar atmospheres part of this course :

$$\frac{N_1^+}{N_1} = \frac{2g_1^+}{N_e g_1} \frac{(2\pi m_e kT)^{3/2}}{h^3} e^{-\chi_{ion}/kT}$$

where  $m_e$  is the mass of the electron,  $\chi_{ion}$  is the ionization energy,  $N_1^+$  and  $N_1$  are the number density of ions and neutral atoms in their ground state,  $N_e$  is the electron number density ,  $g_1^+$  and  $g_1$  are the statistical weight of the ground state of the ion and neutral atom.

- In general, the Saha equation can be used to compute ionization fractions over most of the star. It does, however, require that the gas be in the thermodynamic equilibrium. This is true throughout almost the whole star, as at high densities, collisions will control the level populations. This approximation only breaks down in the solar corona, where the densities become very low.
- However, the Saha equation also breaks down in the centers of stars, where high densities cause the ionization energies of atoms to be reduced. Indeed, if the mean distance between atoms is  $d$ , then there can be no bound states with radii greater than  $\sim d/2$ . In practice, the Saha equation begins to break down at nuclear distances of  $\sim 10a_0$  ( $\sim 10$  Bohr radii).
- To correct for this effect, the Saha equation is normally used until it begins to show decreasing ionization fractions toward the center of the star. When this happens, complete ionization is assumed.

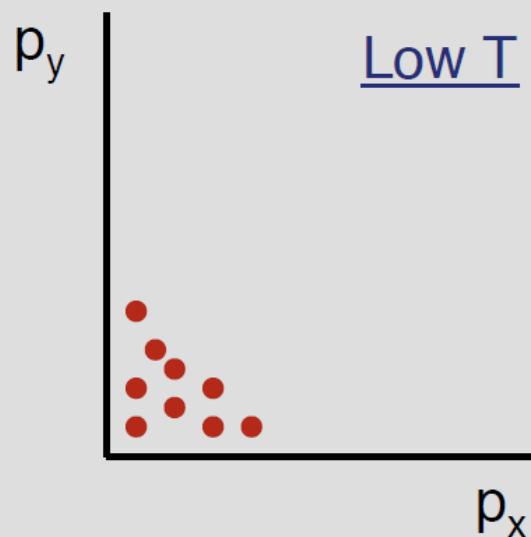
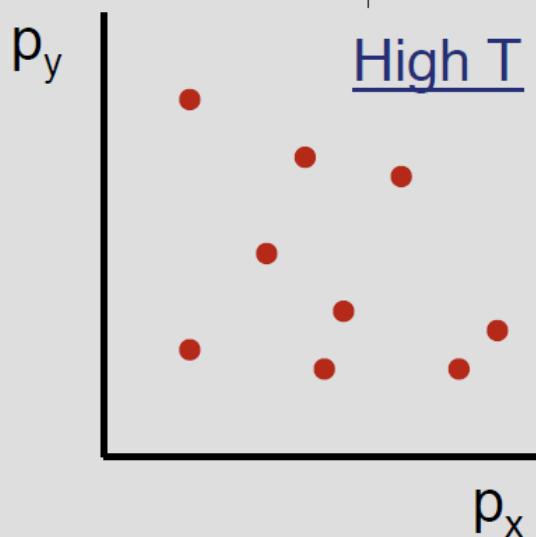
# Degeneracy Pressure (1)

203

- An ordinary classical gas:  $P_{\text{gas}} \propto T \rightarrow 0$  as  $T \rightarrow 0$
- Simultaneously, the mean speed of particles in the gas also goes to zero:

$$v = \sqrt{2kT/m}$$

- The momenta are given by:  $p_x = mv_x$ ;  $p_y = mv_y$ ;  $p_z = mv_z$
- ... if we plot the momenta of particles in a 3D space of  $p_x$ ,  $p_y$  and  $p_z$  then as  $T$  decreases the particles become concentrated near the origin:



# Degeneracy Pressure (2)

204

At low enough temperatures / high enough densities, the concentration of particles with similar (low) momenta would violate the Pauli exclusion principle:

**No two electrons can occupy the same quantum state**

i.e. have the same momentum, spin, and location.

To avoid violating the exclusion principle, electrons in a dense, cold gas **must have larger momenta than we would predict classically.**

Since the pressure  $P$  is mean rate of transport of momentum across unit area

$$P = \frac{1}{3} \int_0^{\infty} v p n(p) dp$$

...where  $n(p)dp$  is the number of particles with momentum between  $p$  and  $p+dp$

... larger momentum means higher pressure.

This quantum mechanical source of pressure is **degeneracy pressure**.

# Degeneracy Pressure (3)

205

This **quantum mechanical** source of pressure is **degeneracy pressure**.

We will discuss it later.

$$P_{ideal} = \frac{k}{\mu m_p} \rho T$$

- **Non-relativistic degeneracy pressure** (speeds  $v \ll c$ ) :

$$P_{deg} = K_1 \rho^{5/3} = K_1 \rho^{1+\frac{1}{n}}$$

← A polytrope of index  $n=1.5$

- $K_1$  is constant
- Does not depend upon temperature for low enough  $T$
- Depends upon composition via the relation between  $N_e$  and  $\rho$

- **Relativistic degeneracy pressure** :

$$P_{deg} = K_2 \rho^{4/3}$$

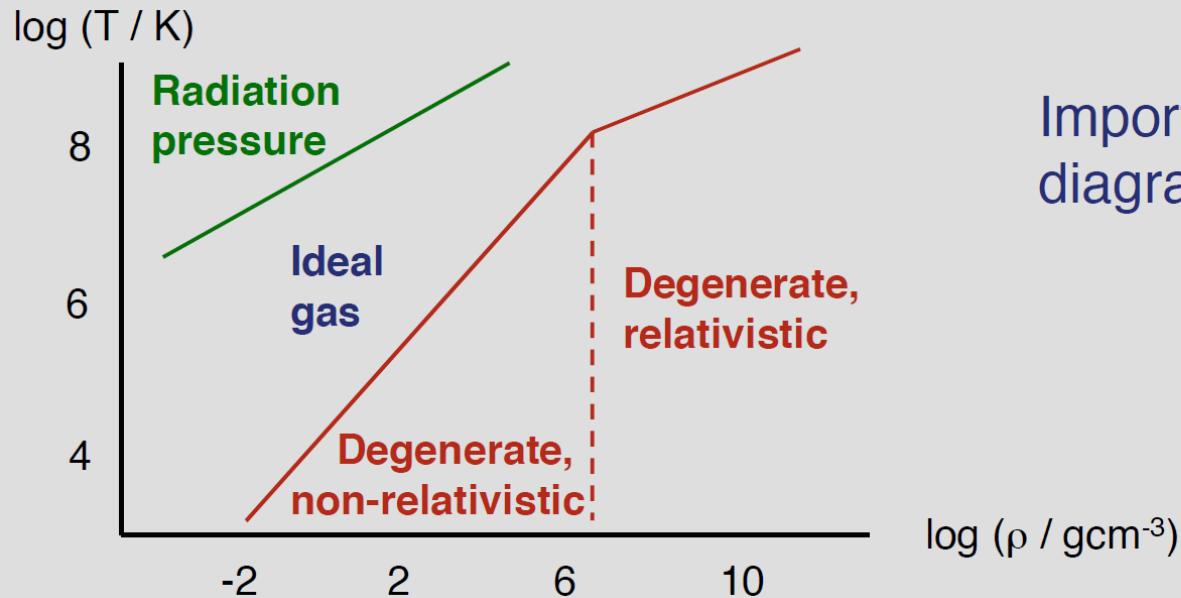
← A polytrope of index  $n=3$

- $K_2$  is another constant
- Equation of state for relativistic degenerate matter, which applies at high density. This is a “softer” equation of state, since  $P$  rises more slowly with increasing density than for the non-relativistic case.

A relation of the form  $P = K \rho^{1+\frac{1}{n}}$  where  $K$  and  $n$  are constants is called a **polytropic relation**, and  $n$  is the **polytropic index**.

# When do the different pressures matter?

206



Different types of star occupy different portions of the plane:

- Solar-type stars - ideal gas throughout
- Massive stars - radiation pressure
- White (and brown) dwarfs - non-relativistic degeneracy pressure

Relativistic degeneracy implies an **unstable** equation of state, so **no stable stars** in that part of the plane.

# Radiation Pressure

207

We have already showed before that radiation pressure can be neglected for Solar-type stars:

$$\frac{P_{rad}}{P_g} = \frac{aT^4}{3} / \frac{kT\rho}{\mu m_p} = \frac{\mu a}{3\mathfrak{R}} \frac{T^3}{\rho} \approx 10^{-4} \text{ (for the Sun)}$$

But becomes **very important** for early-type stars due to the  $T^4$  sensitivity.

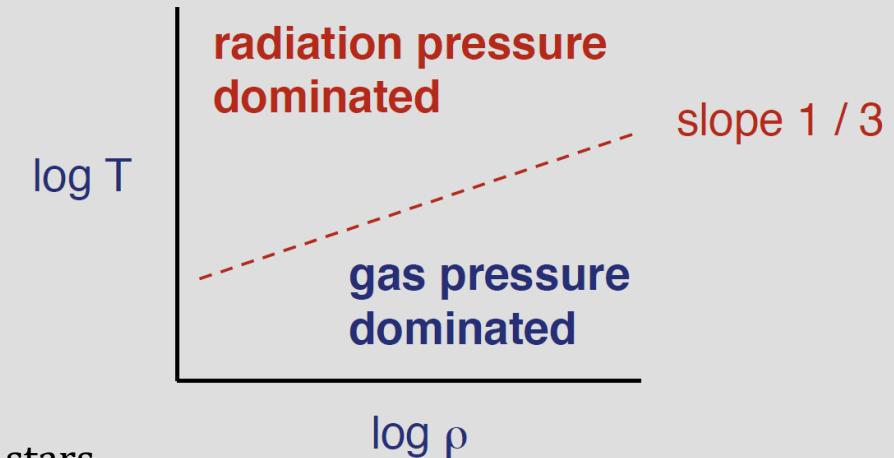
In which stars are gas and radiation pressure important?

$$\left. \begin{aligned} P_{rad} &= \frac{aT^4}{3} \\ P_g &= \frac{\mathfrak{R}T\rho}{\mu} \end{aligned} \right\} \text{equal when } T^3 = \frac{3\mathfrak{R}}{a\mu} \rho$$

From the virial theorem (see Lecture 3):

$$\bar{T} \propto \frac{M}{R} \Rightarrow \frac{P_{rad}}{P_g} \propto M^2$$

i.e.  $P_{rad}$  becomes more significant in higher mass stars.



# Effect of radiation pressure

208

For stars in which radiation pressure plays a non-negligible role we can write the generalized form of the equation of hydrostatic support (Lecture 2):

$$\frac{dP(r)}{dr} + g\rho(r) = a\rho(r)$$

Then

$$\frac{dP(r)}{dr} = -g\rho(r) - \frac{dP_{rad}}{dr} = -g_{\text{eff}}(r)\rho(r)$$

From Lecture 6 (slide 169):

$$\frac{dP_{rad}}{dr} = -\frac{\rho\kappa_R}{c}F \quad \Rightarrow \quad g_{\text{eff}}(r) = g - \frac{\kappa_R}{c}F$$

Consider relative contributions of radiation and (ideal) gas pressures:

$$P_g = \beta P = \frac{\mathfrak{R}T\rho}{\mu}, \quad P_{rad} = (1 - \beta)P = \frac{aT^4}{3}$$

Exclude temperature:  $P = \left[ \left( \frac{\mathfrak{R}}{\mu} \right)^4 \frac{3}{a} \frac{1-\beta}{\beta^4} \right]^{1/3} \rho^{4/3} \quad \Rightarrow \quad P = K\rho^{1+\frac{1}{n}}$  ← A polytrope of index  $n=3$